

# Reversible Conjugation of Ethacrynic Acid with Glutathione and Human Glutathione S-Transferase P1-1<sup>1</sup>

J. H. T. M. Ploemen,<sup>2</sup> A. Van Schanke, B. Van Ommen, and P. J. Van Bladeren

Department of Biological Toxicology, TNO Toxicology and Nutrition Institute, P. O. Box 360, 3700 AJ Zeist, the Netherlands

## ABSTRACT

The reversibility of the conjugation reaction of the diuretic drug ethacrynic acid (EA), an  $\alpha,\beta$ -unsaturated ketone, with glutathione and glutathione S-transferase P1-1 (GST P1-1) has been studied. When the glutathione conjugate of EA was incubated with a 5-fold molar excess of *N*-acetyl-L-cysteine or GST P1-1, a time-dependent transfer of EA to *N*-acetyl-L-cysteine or GST P1-1 was observed. With increasing pH, the pseudo first order rate constants of transfer of EA to *N*-acetyl-L-cysteine increased from 0.010 h<sup>-1</sup> (pH 6.4) to 0.040 h<sup>-1</sup> (pH 7.4) and 0.076 h<sup>-1</sup> (pH 8.4).

From the fact that preincubation of GST P1-1 with 1-chloro-2,4-dinitrobenzene reduced the incorporation of [<sup>14</sup>C]EA from 0.94 ± 0.21 (SD) to 0.16 ± 0.02 mol EA/mol subunit and from automated Edman degradation of the major radioactive peptide isolated after pepsin digestion of the [<sup>14</sup>C]EA-labeled enzyme, it was concluded that the reaction of EA takes place with cysteine 47 of GST P1-1.

When GST P1-1 was inactivated with a 5-fold molar excess of EA, adding an excess of glutathione resulted in full restoration of the catalytic activity in about 120 h.

These findings may have several implications. Under normal physiological conditions the inhibition of GST P1-1 by covalent binding of EA would be reversed by glutathione, leaving reversible inhibition by the glutathione conjugate of EA and by EA itself as the main mechanism of inhibition; however, when glutathione levels are low the covalent inhibition might be predominant, resulting in a completely different time course for the inhibition.

## INTRODUCTION

Conjugation with the tripeptide glutathione is considered to be an important detoxification reaction for electrophilic xenobiotics, including numerous cytostatic agents. In general this reaction is catalyzed by GST<sup>3</sup> (1-3). Most GSTs occur in cytosol and belong to one of four multigene families, termed alpha, mu, pi, and theta (1, 4).

The effectiveness of many clinically useful anticancer drugs is severely limited by drug resistance. Several findings suggest the involvement of GST, besides other mechanisms, especially with respect to resistance to alkylating agents such as chlorambucil and melphalan (5-7). These findings include GST overexpression of especially the pi class in tumors (8, 9), the direct conjugation of alkylating agents by GST (10, 11), and the overexpression of GST in yeast and mammalian cell lines by genetic engineering which confers to these organisms resistance to alkylating agents (6, 12). A promising strategy to overcome this alkylator resistance phenotype may be based on inhibition of GST.

The diuretic drug ethacrynic acid, an  $\alpha,\beta$ -unsaturated ketone, is a potent reversible inhibitor of GST isoenzymes (13-15) and has been used to study the role of GST in drug resistance *in vitro*, using cell

lines (16), and colon tumor xenografts (17) and in a phase I clinical study with the cytostatic agent thiotepa (18). Moreover, a concentration-dependent inhibition by ethacrynic acid of the enzyme-catalyzed conjugation of glutathione with the clinically important alkylating agent chlorambucil has been reported (11). The reversible inhibition would further be enhanced by the formation of the glutathione conjugate of ethacrynic acid, which is an even stronger inhibitor for all GSTs but the pi class (14). For both human and rat GST of the pi class, covalent modification of GST concomitant with an irreversible loss of activity could be achieved using slightly more drastic incubation conditions (19). Conjugation with glutathione does not always lead to the detoxification of electrophilic xenobiotics (20). In addition to glutathione conjugates that are reactive by themselves, other types of glutathione conjugates may undergo further metabolism to a reactive species (20). A special case involves glutathione conjugates that exert their toxic effects through release of reactive species; the glutathione conjugates serve as transport and targeting agents. This situation occurs when the glutathione conjugation reaction is reversible, as found, *e.g.*, for some methyl isocyanates (21) and isothiocyanates (22). The Michael addition of glutathione with  $\alpha,\beta$ -unsaturated aldehydes and ketones is another well known reversible reaction (23). A reversible Michael reaction has, *e.g.*, been shown to be involved in the covalent of an  $\alpha,\beta$ -unsaturated ketone metabolite of the veterinary drug furazolidone to glutathione and protein (24).

Since ethacrynic acid also contains an  $\alpha,\beta$ -unsaturated ketone moiety, the present study was designed to investigate the reversible covalent interaction of ethacrynic acid with glutathione as well as with GST P1-1. The interaction of ethacrynic acid with this enzyme was included, since the inactivation of the GST of the pi class in several cases has been shown to be the result of the modification of a highly reactive cysteine residue (25, 26) and since pi class, next to alpha, is one of the primary GST classes that are involved in drug resistance (5).

## MATERIALS AND METHODS

**Chemicals and Enzymes.** Ethacrynic acid [2,3-dichloro-4-(2-methylene-1-oxobutyl)phenoxy]acetic acid, S-hexylglutathione, *N*-acetyl-L-cysteine, and Tris were from Sigma Chemical Co., St. Louis, MO. Epoxy-activated Sepharose 6B was purchased from Pharmacia, Uppsala, Sweden. [<sup>14</sup>C]Ethacrynic acid (15 mCi/mmol) was purchased from Amersham, Buckinghamshire, United Kingdom. Trifluoroacetic acid was from J. T. Baker Chemical Co., Inc., Philipsburg, NJ. Pepsin (from porcine gastric mucosa) was obtained from Boehringer Mannheim, Mannheim, Germany.

The radioactive conjugate of ethacrynic acid was prepared by adding 6  $\mu$ mol of glutathione in 180  $\mu$ l of 0.1 M potassium phosphate buffer, pH 8, with 50% ethanol, to 1.3  $\mu$ mol of [<sup>14</sup>C]ethacrynic acid. After overnight incubation, the glutathione conjugate of ethacrynic acid was purified by preparative RP-HPLC using Zorbax ODS (Dupont; 21.2 \* 250 mm), eluted at a flow rate of 4 ml/min with 0.01% formic acid (solvent I) and methanol (solvent II), with a linear gradient of 40-100% solvent II in 60 min (*k'* = 2.0 and 3.1 for the conjugate and ethacrynic acid, respectively). About 70% conversion of ethacrynic acid to the glutathione conjugate was obtained. Methanol was removed under N<sub>2</sub>, after which a stock solution (of 136  $\mu$ M) of the glutathione conjugate

Received 9/20/93; accepted 12/16/93.

The costs of publication of this article were defrayed in part by the payment of page charges. This article must therefore be hereby marked advertisement in accordance with 18 U.S.C. Section 1734 solely to indicate this fact.

<sup>1</sup> This investigation was supported by the NWO-Foundation for Medical and Health Research MEDIGON (Grant 900-521-124).

<sup>2</sup> To whom requests for reprints should be addressed.

<sup>3</sup> The abbreviations used are: GST, glutathione S-transferase; CDNB, 1-chloro-2,4-dinitrobenzene; HPLC, high-performance liquid chromatography; RP, reverse phase; dd, doublet of doublets; d, doublet; s, singlet; m, multiplet; dm, doublet of multiplets.

was stored at  $-30^{\circ}\text{C}$ . A product of 95+% purity was obtained as judged with RP-HPLC analysis, with a retention time identical to that of the nonradioactive conjugate (14).

The *N*-acetyl-L-cysteine conjugate of ethacrynic acid was prepared, in analogy to the synthesis of the glutathione conjugate (14). The  $^1\text{H}$  nuclear magnetic resonance (400 MHz,  $\text{D}_2\text{O}$ ) spectra of the *N*-acetyl-L-cysteine conjugate are consistent with the expected structure (19), with the following information for the ethacrynic acid part of  $\delta$  7.81/7.79 (dd, 1H,  $J = 8.8$  Hz),  $\delta$  7.23 (d, 1H,  $J = 8.8$  Hz), 4.99 (s, 2H),  $\delta$  3.8 (m, 1H),  $\delta$  1.95/1.81 (dm, 2H),  $\delta$  1.1 (m, 3H); and for the *N*-acetyl-L-cysteine part of  $\delta$  2.19 (s, 3H,  $-\text{CH}_3$ ). The proton signals of cys  $\alpha$  overlap with  $\text{D}_2\text{O}$  ( $\delta = 4.7$ ), while the signals of cys  $\beta$  and the protons next to the sulfur atom (of the ethacrynic acid moiety) were found in the region of  $\delta$  2.9–3.1, as a complicated multiplet pattern.

GST P1-1 was purified from human placenta as described (19). Protein was determined by the method of Lowry *et al.* (27), using bovine serum albumin as standard.

**Incubations.** The glutathione conjugate of ethacrynic acid (0.5 mM) was incubated at  $20^{\circ}\text{C}$  with *N*-acetyl-L-cysteine (2.5 mM) in 0.4 ml of 0.1 M potassium phosphate buffer with 0.1 mM EDTA at three pH levels (6.4, 7.4, and 8.4). For each pH, 22 independent samples were prepared. At each time point, 20  $\mu\text{l}$  were injected on RP-HPLC, using a Zorbax ODS (250  $\times$  4.6 mm; LC Service, the Netherlands) column and eluted at a flow rate of 1 ml/min with 0.1% trifluoroacetic acid in deionized water (solvent A) and in methanol (solvent B), with a linear gradient of 30–95% solvent B in 18 min, followed by 2 min at 95% solvent B ( $k' = 4.5, 5.1, \text{ and } 6.0$  for the glutathione conjugate, *N*-acetyl-L-cysteine conjugate, and ethacrynic acid, respectively). Peak areas at 270 nm were integrated with Nelson Analytical Model 2600 Chromatography Software. EDTA was added to the incubations to prevent the trapping agent *N*-acetyl-L-cysteine and the free glutathione from oxidation to their disulfides. Since no free ethacrynic acid could be detected, it is concluded that EDTA protected sufficiently against oxidation.

Covalent binding of [ $^{14}\text{C}$ ]ethacrynic acid was studied in a volume of 75  $\mu\text{l}$  0.1 M potassium phosphate buffer (pH 7.4) with 0.1 mM EDTA (buffer A), after preincubation of 25  $\mu\text{M}$  GST P1-1 for 75 min at room temperature with ( $n = 3$ ) or without ( $n = 2$ ) 1 mM CDNB, whereafter the enzyme was incubated for 110 min with [ $^{14}\text{C}$ ]ethacrynic acid (final concentration, 100  $\mu\text{M}$ ). Enzyme-bound ethacrynic acid was separated from ethacrynic acid by RP-HPLC (Vydac TP5 column; Chrompack, the Netherlands; 200  $\times$  3 mm). Elution was performed with a flow of 0.6 ml/min, with solvent A (see above) and 0.1% trifluoroacetic acid in acetonitrile (solvent C), with a linear gradient of 30–60% solvent C in 30 min ( $k' = 4.0, \text{ and } 6.3$  for ethacrynic acid and enzyme with bound ethacrynic acid, respectively). UV detection (at 214 nm) was used to identify the enzyme peak, while simultaneously the radioactivity was measured using an on-line radiochemical detector.

The [ $^{14}\text{C}$ ]ethacrynic acid-labeled GST P1-1 (0.25 mg) was digested with pepsin [enzyme/protein ratio, 1/20 (w/w)] in 0.05 M Tris/ $\text{H}_3\text{PO}_4$  (pH 1.8) for 18 h at  $37^{\circ}\text{C}$ . The pepsin-peptide mixture was purified on a RP column ( $\text{C}_{18}$ , Vydac Protein & Peptide USA; 250  $\times$  4.6 mm) and eluted with solvents A and C (see above), 5 min isocratically at 100% solvent A, followed by a linear gradient from 0–60% solvent C in 70 min (flow rate, 1 ml/min). The main radioactive peak was repeatedly purified on the same column. The peptide was degraded using automated Edman degradation on an Applied Biosystems Model 475 peptide sequencer on-line connected to a Model 120A PTH analyzer.

The catalytic activity of GST P1-1, inactivated with ethacrynic acid, was monitored after the addition of glutathione. GST P1-1 (1  $\mu\text{M}$ ) was preincubated in buffer A (see above) with or without 10  $\mu\text{M}$  ethacrynic acid (final volume, 200  $\mu\text{l}$ ), after which glutathione was added (final concentrations, 0, 10, 100, and 1000  $\mu\text{M}$ ). These incubations were performed in triplicate, at room temperature. At various time points, 20-pmol enzyme samples were transferred to cuvettes, after which the catalytic activity towards CDNB was measured (28). A time series was stopped when the remaining catalytic activity in the corresponding blank incubation was less than 70%.

To study the interaction of GST P1-1 with the glutathione conjugate of [ $^{14}\text{C}$ ]ethacrynic acid, seven independent samples of 10  $\mu\text{M}$  GST P1-1 were incubated at room temperature with 2  $\mu\text{M}$  radioactive glutathione conjugate in buffer A (see above) (final volume, 50  $\mu\text{l}$ ). To separate the glutathione con-

jugate and free ethacrynic acid from the enzyme with bound ethacrynic acid, 30  $\mu\text{l}$  were injected on the Vydac TP5 column (see above).

## RESULTS

The occurrence of the retro Michael cleavage of the glutathione conjugate of ethacrynic acid was studied by incubation of the glutathione conjugate of ethacrynic acid with an excess of *N*-acetyl-L-cysteine (Fig. 1). The transfer of the ethacrynic acid moiety to *N*-acetyl-L-cysteine was followed with time by quantification of the glutathione and *N*-acetyl-L-cysteine conjugates of ethacrynic acid on RP-HPLC. The rate of transfer increased with increasing pH, with pseudo first order rates of 0.010, 0.040, and 0.076  $\text{h}^{-1}$  for pH 6.4, 7.4, and 8.4, respectively. After 180 h of incubation at pH 8.4, an equilibrium was reached between the glutathione and the *N*-acetyl-L-cysteine conjugate, suggesting that the distribution of ethacrynic acid over glutathione and *N*-acetyl-L-cysteine is mainly determined by their relative concentrations.

Retro Michael cleavage can also occur with GST P1-1-bound ethacrynic acid, if the assumption is right that ethacrynic acid reacts with a cysteine residue of GST P1-1 (19). In order to check this assumption, the incorporation of [ $^{14}\text{C}$ ]ethacrynic acid in GST P1-1 after preincubation with CDNB was studied, which is known to inactivate GST P1-1 by modification of cysteine 47 (25). Identical to an earlier study (19),  $0.94 \pm 0.21$  nmol label/nmol GST P1-1 could be incorporated in blank incubations. As expected, CDNB protects against incorporation of ethacrynic acid;  $0.16 \pm 0.02$  (SD) nmol label/nmol GST P1-1 could be incorporated after preincubation with CDNB, supporting the hypothesis that ethacrynic acid reacts with cysteine 47 of GST P1-1. In order to definitely identify the amino acid involved in the reaction, the GST P1-1 with bound ethacrynic acid was digested with pepsin and the resulting peptides were separated on HPLC. A main radioactive peak was identified, which contained >80% of the radioactivity, eluting at 45 min (Fig. 2). The amino acid sequence of this peptide indicated that it spans residues 44–46 in the primary amino acid sequence of GST P1-1 (Lys–Ala–Ser) (29–31), while an unknown residue was observed in the 4th cycle (presumably the cysteine-ethacrynic acid adduct). Thus it was again concluded that cysteine 47 is the main target site.

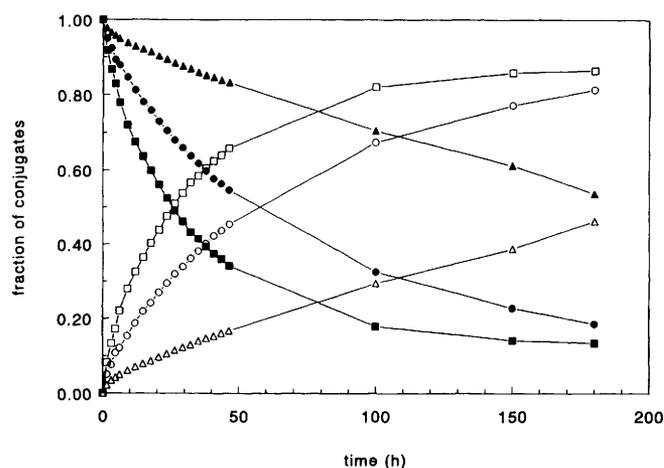
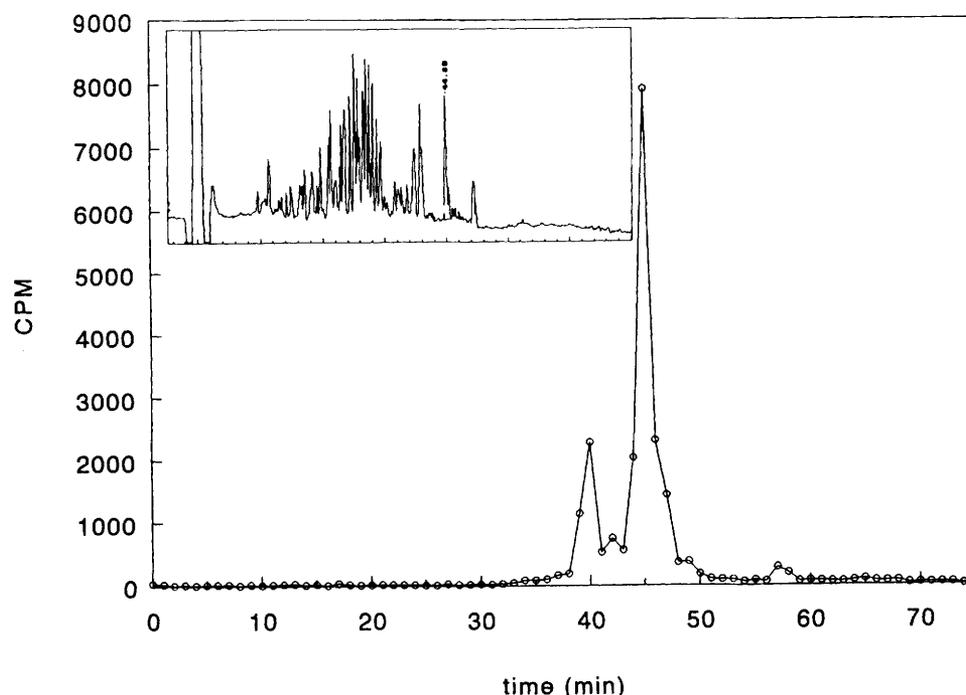


Fig. 1. Transfer of ethacrynic acid from its glutathione conjugate to *N*-acetyl-L-cysteine. The glutathione conjugate of ethacrynic acid (0.5 mM) was incubated at  $20^{\circ}\text{C}$  with a 5-fold excess of *N*-acetyl-L-cysteine. The transfer of the ethacrynic acid moiety to *N*-acetyl-L-cysteine was followed with time by quantification of the conjugates on RP-HPLC by integration of the peak areas at 270 nm. The reaction was studied at three pH levels, 6.4 ( $\blacktriangle, \triangle$ ), 7.4 ( $\bullet, \circ$ ), and 8.4 ( $\blacksquare, \square$ ). Open symbols, newly formed *N*-acetyl-L-cysteine conjugates; closed symbols, the glutathione conjugate of ethacrynic acid.

Fig. 2. HPLC analysis of the pepsin digest of ethacrynic acid-labeled GST P1-1. From 1-min fractions, samples (40  $\mu$ l) were screened for radioactivity. *Inset*, peptides monitored from 0 to 75 min at 214 nm; peak at 44.8 min indicated (full scale, 0.3 absorbance unit full scale).



Then, GST P1-1 (1  $\mu$ M) was incubated with ethacrynic acid (10  $\mu$ M), resulting in 90% loss of activity towards CDNB, and glutathione was added. The catalytic activity toward CDNB was measured over a 120-h period (Fig. 3). Full restoration of the catalytic activity occurs

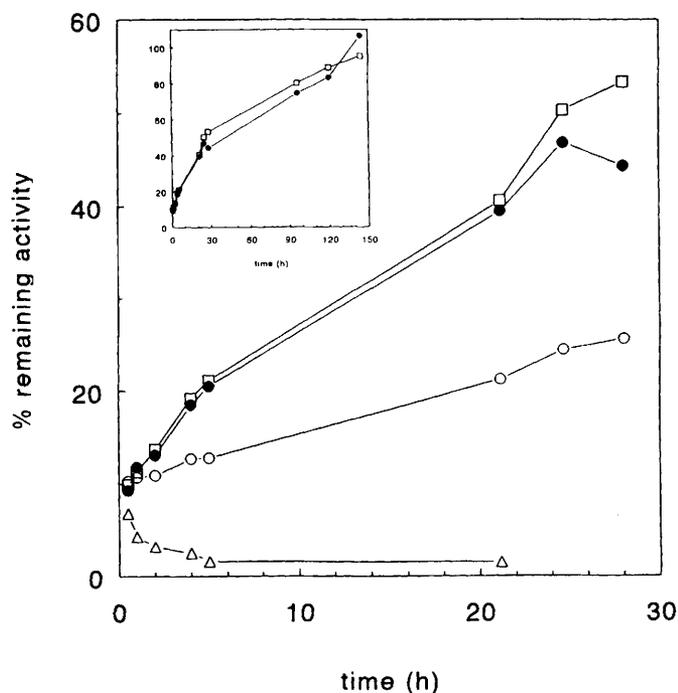


Fig. 3. Restoration of the catalytic activity of ethacrynic acid-inactivated human GST P1-1 by incubation with glutathione. GST P1-1 (1  $\mu$ M) was incubated with ethacrynic acid (10  $\mu$ M), resulting in 90% loss of catalytic activity towards CDNB. Glutathione was added [0 ( $\Delta$ ), 0.01 ( $\circ$ ), 0.1 ( $\square$ ), and 1 ( $\bullet$ ) mM]. The catalytic activity toward CDNB was measured over a 120-h period and expressed as a percentage of control incubations with non-ethacrynic acid-modified enzyme (note: incubations were stopped when the remaining activity in the corresponding control was less than 70%). *Inset*, effect of prolonged incubation. Individual points are the average of three measurements, with coefficients of variation less than 15%.

with 0.1 and 1 mM glutathione (Fig. 3, *inset*). The 10  $\mu$ M incubation initially shows partial restoration of catalytic activity, but after prolonged incubation loss of catalytic activity is observed, probably due to oxidation. This is also observed in the corresponding control incubation (a 30% loss of activity in about 30 h; result not shown). Without a trapping agent for free ethacrynic acid, no restoration of activity was observed (Fig. 3).

In order to investigate whether retro Michael cleavage of the glutathione conjugate of ethacrynic acid concomitant with incorporation of ethacrynic acid in GST P1-1 occurs, a 5-fold molar excess of the enzyme was incubated with the glutathione conjugate of [ $^{14}$ C]ethacrynic acid. A time-dependent increase of enzyme-bound label was observed (Fig. 4), in accordance with the reversible nature of the reactions.

## DISCUSSION

$\alpha,\beta$ -unsaturated aldehydes and ketones have long been known to form conjugates with glutathione, both spontaneously and enzyme catalyzed (32). The extent to which the enzyme plays a role differs widely among members of this class of compounds (33). The chemical reaction involved in the conjugation of ethacrynic acid and structurally related compounds, a Michael addition, is reversible. In the present study, it was shown that this retro Michael cleavage of ethacrynic acid and glutathione indeed occurs. Thus, ethacrynic acid may be transferred from one low molecular weight compound to another or to reactive and accessible cysteines in proteins, e.g., cysteine 47 of GST P1-1 as observed. Transfer from the glutathione conjugate of ethacrynic acid might take place in one of two ways, namely a release from the glutathione conjugate in the active site of the enzyme and/or reconjugation of released ethacrynic acid from the free glutathione conjugate in the incubation medium, followed by binding to the enzyme. This transfer phenomenon is probably a common feature of  $\alpha,\beta$ -unsaturated aldehydes and ketones; transport via a thiol conjugate and subsequent regeneration of the reactive agent thus may be involved in the biological activity of such adducts (23).

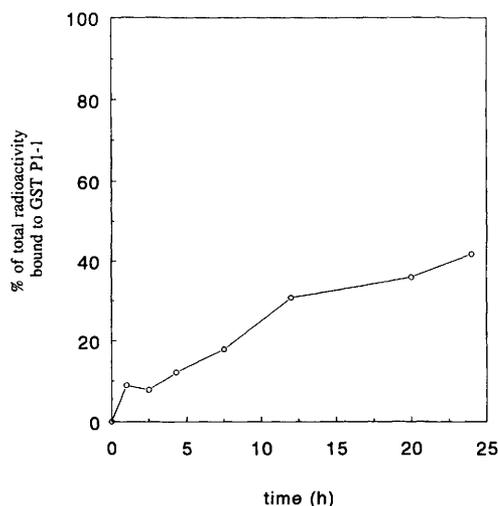


Fig. 4. Interaction of GST P1-1 with the glutathione conjugate of [ $^{14}$ C]ethacrynic acid. GST P1-1 (10  $\mu$ M) was incubated with a 2  $\mu$ M concentration of the glutathione conjugate,  $^{14}$ C-labeled in the ethacrynic acid moiety. To separate the glutathione conjugate and free ethacrynic acid from the enzyme-bound fraction, a RP-HPLC method was used as described under "Materials and Methods." The transfer of the ethacrynic acid moiety is expressed as a percentage of total  $^{14}$ C label.

The nature of the inhibition of GST by ethacrynic acid has been studied in detail, since it was reported that ethacrynic acid *in vivo* bound covalently to rat GST 3-4 (34). Previously we showed that ethacrynic acid and its glutathione conjugate were potent reversible inhibitors of GST isoenzymes, with 50% inhibitory values in the range of 1–10  $\mu$ M (14). This reversible inhibition was suggested to be the predominant inhibitory mechanism *in vivo*, since the incorporation of ethacrynic acid in GST 3-4 did not appear to inactivate the enzyme (14, 19). In the case of GST P1-1, it was shown recently that the mechanisms of the reversible inhibition by ethacrynic acid and its glutathione conjugate were distinct (noncompetitive and competitive, respectively) (35). In the alpha and mu class the conjugate of ethacrynic acid was a more potent reversible inhibitor than the parent compound, while for the pi class some conflicting results have been reported (14, 35).

GSTs of the pi class are inhibited by covalent binding of  $\alpha,\beta$ -unsaturated aldehydes and ketones; acrolein, a toxic aldehyde that occurs as an environmental pollutant (36), and also ethacrynic acid (19), specifically inactivated GST of the pi class. It is now clear that the inhibition is only transitory, since the chemical reaction is reversible. Full restoration of the catalytic activity can be achieved by prolonged incubation with an excess of glutathione. In an earlier study, only marginal inactivation of GST 7-7 by ethacrynic acid was observed using overnight dialysis experiments (14). Presumably the reversibility of the chemical reaction contributed to the failure of this dialysis experiment to detect time-dependent inhibition of GST P1-1 by ethacrynic acid.

The implications of the present findings may be several. Firstly, this mode of action, reversible covalent binding to GST P1-1 and glutathione, may have some significance for the use of ethacrynic acid as an *in vivo* inhibitor for GST P1-1 in drug resistance. Under normal physiological conditions [glutathione concentration 1–10 mM (1)], glutathione may be expected to reverse any covalent binding of ethacrynic acid to GST P1-1, and the inhibition of GST would occur only reversibly, through the glutathione conjugate of ethacrynic acid and of ethacrynic acid itself. However, in those cells with high levels of GST P1-1 and/or low levels of glutathione, covalent inhibition of GST P1-1 might be predominant. The time course of inhibition would be completely different in these two cases.

Recently, it was shown that chronic exposure of human colon carcinoma cells to ethacrynic acid led to a 2–3-fold increase of GST P1-1 activity, by an induction of the enzyme at the transcriptional level (37). This phenomenon has been proposed by Talalay *et al.* (38) to be a general one; compounds that contain a Michael acceptor or from which a Michael acceptor can be formed during metabolism are usually inducing agents for GST. The contrasting effects observed for Michael acceptors, *i.e.*, inhibition of GST by covalent modification and induction of GST, deserve further attention.

Furthermore, some  $\alpha,\beta$ -unsaturated aldehydes are established inhibitors of growth. In the case of 4-hydroxynonenal and related compounds it was shown that inhibition of DNA synthesis was involved, presumably as a result of a reaction with a functional sulfhydryl group of DNA polymerase (23). More recently, another type of growth inhibition has been reported, which involves the  $\alpha,\beta$ -unsaturated ketone  $\Delta^{12}$  prostaglandin  $J_2$ , a cyclopentenone prostaglandin which readily forms glutathione conjugates (39). These conjugations should in principle also be able to undergo retro Michael cleavage to reform the parent compounds. Interestingly, it has been shown that ethacrynic acid, along with other inhibitors of GST, also has antiproliferative effects on cell lines (40), which seem to be reversible.

## REFERENCES

- Mannervik, B. The isoenzymes of glutathione transferase. *Adv. Enzymol. Relat. Areas Mol. Biol.*, 57: 357–417, 1985.
- Mannervik, B., and Danielson, U. H. Glutathione transferases—structure and catalytic activity. *CRC Crit. Rev. Biochem.*, 23: 283–337, 1988.
- Armstrong, R. N. Glutathione *S*-transferases: reaction mechanism, structure, and function. *Chem. Res. Toxicol.*, 4: 131–140, 1991.
- Meyer, D. J., Coles, B., Pemble, S. E., Gilmore, K. S., Fraser, G. M., and Ketterer, B. Theta, a new class of glutathione transferases purified from rat and man. *Biochem. J.*, 274: 409–414, 1991.
- Waxman, D. J. Glutathione *S*-transferases: role in alkylating agent resistance and possible target for modulation—a review. *Cancer Res.*, 50: 6449–6454, 1990.
- Black, S. M., and Wolf, C. R. The role of glutathione-dependent enzymes in drug resistance. *Pharmacol. Ther.*, 51: 139–154, 1991.
- Morrow, C. S., and Cowan, K. H. Glutathione *S*-transferases and drug resistance. *Cancer Cells*, 2: 15–22, 1990.
- Wang, A. L., and Tew, K. D. Increased glutathione *S*-transferase activity in a cell line with acquired resistance to nitrogen mustards. *Cancer Treat. Rep.*, 69: 677–682, 1985.
- Sato, K. Glutathione transferases as markers of preneoplasia and neoplasia. *Adv. Cancer Res.*, 52: 205–255, 1989.
- Dulik, D. M., Fenselau, C., and Hilton, J. Characterization of melphalan-glutathione adducts whose formation is catalysed by glutathione transferases. *Biochem. Pharmacol.*, 35: 3405–3409, 1986.
- Ciaccio, P. J., Tew, K. D., and LaCreta, F. P. Enzymatic conjugation of chlorambucil with glutathione by human glutathione *S*-transferases and inhibition by ethacrynic acid. *Biochem. Pharmacol.*, 42: 1504–1507, 1991.
- Lewis, A. D., Hickson, I. D., Robson, C. N., Harris, A. L., Hayes, J. D., Griffiths, S. A., Manson, M. M., Hall, A. E., Moss, J. E., and Wolf, C. R. Amplification and increased expression of alpha class glutathione *S*-transferase-encoding genes associated with resistance to nitrogen mustards. *Proc. Natl. Acad. Sci. USA*, 85: 8511–8515, 1988.
- Ahokas, J. T., Nicholls, F. A., Ravenscroft, P. J., and Emmerson, B. T. Inhibition of purified rat liver glutathione *S*-transferase isoenzymes by diuretic drugs. *Biochem. Pharmacol.*, 34: 2157–2161, 1985.
- Ploemen, J. H. T. M., Van Ommen, B., and Van Bladeren, P. J. Inhibition of rat and human glutathione *S*-transferase isoenzymes by ethacrynic acid and its glutathione conjugate. *Biochem. Pharmacol.*, 40: 1631–1635, 1990.
- Phillips, M. F., and Mantle, T. J. The initial-rate kinetics of mouse glutathione *S*-transferase YfYf. Evidence for an allosteric site for ethacrynic acid. *Biochem. J.*, 275: 703–709, 1991.
- Tew, K. D., Bomber, A. M., and Hoffman, S. J. Ethacrynic acid and piriprost as enhancers of cytotoxicity in drug resistance and sensitive cell lines. *Cancer Res.*, 48: 3622–3625, 1988.
- Clapper, M. L., Hoffman, S. J., and Tew, K. D. Sensitization of human colon tumor xenografts to *l*-phenylalanine using ethacrynic acid. *J. Cell Pharmacol.*, 1: 71–78, 1990.
- O'Dwyer, P. J., LaCreta, F., Nash, S., Tinsley, P. W., Schilder, R., Clapper, M. L., Tew, K. D., Panting, L., Litwin, S., Comis, R. L., and Ozols, R. F. Phase I study of thiotepa in combination with the glutathione transferase inhibitor ethacrynic acid. *Cancer Res.*, 51: 6059–6065, 1991.
- Ploemen, J. H. T. M., Bogaards, J. J. P., Veldink, G. A., Van Ommen, B., Jansen, D. H. M., and Van Bladeren, P. J. Isoenzyme selective irreversible inhibition of rat and human glutathione *S*-transferases by ethacrynic acid and two brominated derivatives. *Biochem. Pharmacol.*, 45: 633–639, 1993.

20. Monks, T. J., Anders, M. W., Dekant, W., Stevens, J. L., Lau, S. S., and Van Bladeren, P. J. Glutathione conjugate mediated toxicities. *Toxicol. Appl. Pharmacol.*, *106*: 1–19, 1990.
21. Baillie, T. A., and Slatter, J. G. Glutathione: a vehicle for the transport of chemically reactive metabolites *in vivo*. *Acc. Chem. Res.*, *24*: 264–270, 1991.
22. Bruggeman, I. M., Temmink, J. H. M., and Van Bladeren, P. J. Glutathione- and cysteine-mediated cytotoxicity of allyl and benzyl isothiocyanate. *Toxicol. Appl. Pharmacol.*, *83*: 349–359, 1986.
23. Witz, G. Biological interactions of  $\alpha,\beta$ -unsaturated aldehydes. *J. Free Radicals Biol. & Med.*, *7*: 333–349, 1989.
24. Vroomen, L. H. M., Berghmans, M. C. J., Groten, J. P., Koeman, J. H., and Van Bladeren, P. J. Reversible interaction of a reactive intermediate derived from furazolidone with glutathione and protein. *Toxicol. Appl. Pharmacol.*, *93*: 53–60, 1988.
25. Caccuri, A. M., Petruzzelli, R., Polizio, F., Federici, G., and Desideri, A. Inhibition of glutathione transferase  $\pi$  from human placenta by 1-chloro-2,4-dinitrobenzene occurs because of covalent reaction with cysteine 47. *Arch. Biochem. Biophys.*, *297*: 119–122, 1992.
26. Tamai, K., Satoh, K., Tsuchida, S., Hatayama, I., Maki, T., and Sato, K. Specific inactivation of glutathione *S*-transferases in class  $\pi$  by SH-modifiers. *Biochem. Biophys. Res. Commun.*, *167*: 331–338, 1990.
27. Lowry, O. H., Rosebrough, N. J., Farr, A. L., and Randall, R. J. Protein measurement with the Folin phenol reagent. *J. Biol. Chem.*, *193*: 265–275, 1951.
28. Habig, W. H., Pabst, M. J., and Jakoby, W. B. Glutathione *S*-transferases. The first step in mercapturic acid formation. *J. Biol. Chem.*, *249*: 7130–7139, 1974.
29. Kano, T., Sakai, M., and Muramutsa, M. Structure and expression of a human  $\pi$  glutathione *S*-transferase messenger RNA. *Cancer Res.*, *47*: 5626–5636, 1987.
30. Ålin, P., Mannervik, B., and Jörnvall, H. Structural evidence for three different types of glutathione transferase in human tissues. *FEBS Lett.*, *182*: 319–329, 1985.
31. Dao, D. D., Partridge, C. A., Kurosky, A., and Awasthi, Y. C. Human glutathione *S*-transferases. Characterization of the anionic forms from lung and placenta. *Biochem. J.*, *221*: 33–43, 1984.
32. Boyland, E., and Chasseaud, L. F. Enzymes catalysing conjugations of glutathione with  $\alpha,\beta$ -unsaturated carbonyl compounds. *Biochem. J.*, *109*: 651–661, 1968.
33. Ålin, P., Danielson, H., and Mannervik, B. 4-Hydroxyalk-2-enals are substrates for glutathione transferase. *FEBS Lett.*, *197*: 267–270, 1985.
34. Yamada, T., and Kaplowitz, N. Binding of ethacrynic acid to hepatic glutathione *S*-transferases *in vivo* in the rat. *Biochem. Pharmacol.*, *29*: 1205–1208, 1980.
35. Awasthi, S., Srivastava, S. K., Ahmad, F., Ahmad, H., and Ansari, G. A. S. Interactions of glutathione *S*-transferase- $\pi$  with ethacrynic acid and its glutathione conjugate. *Biochim. Biophys. Acta*, *1164*: 173–178, 1993.
36. Berhane, K., and Mannervik, B. Inactivation of the genotoxic aldehyde acrolein by human glutathione transferases of classes alpha, mu, and pi. *Mol. Pharmacol.*, *37*: 251–257, 1990.
37. Kuzmich, S., Vanderveer, L. A., Walsh, E. S., LaCreta, F. P., and Tew, K. D. Increased levels of glutathione *S*-transferase  $\pi$  transcript as a mechanism of resistance to ethacrynic acid. *Biochem. J.*, *281*: 219–224, 1992.
38. Talalay, P., De Long, M. J., and Prochaska, H. J. Identification of a common chemical signal regulating the induction of enzymes that protect against chemical carcinogenesis. *Proc. Natl. Acad. Sci. USA*, *85*: 8261–8265.
39. Koizumi, T., Negeshi, M., and Ichikawa, A. Inhibitory effect of intracellular glutathione on  $\Delta^{12}$ -prostaglandin  $J_2$ -induced protein syntheses in porcine aortic endothelial cells. *Biochem. Pharmacol.*, *44*: 1597–1602, 1992.
40. Sato, Y., Fujii, S., Fujii, Y., and Kaneko, T. Antiproliferative effects of glutathione *S*-transferase inhibitors on the K562 cell line. *Biochem. Pharmacol.*, *39*: 1263–1266, 1990.

# Cancer Research

The Journal of Cancer Research (1916–1930) | The American Journal of Cancer (1931–1940)

## Reversible Conjugation of Ethacrynic Acid with Glutathione and Human Glutathione S-Transferase P1-1

J. H. T. M. Ploemen, A. Van Schanke, B. Van Ommen, et al.

*Cancer Res* 1994;54:915-919.

**Updated version** Access the most recent version of this article at:  
<http://cancerres.aacrjournals.org/content/54/4/915>

**E-mail alerts** [Sign up to receive free email-alerts](#) related to this article or journal.

**Reprints and Subscriptions** To order reprints of this article or to subscribe to the journal, contact the AACR Publications Department at [pubs@aacr.org](mailto:pubs@aacr.org).

**Permissions** To request permission to re-use all or part of this article, use this link  
<http://cancerres.aacrjournals.org/content/54/4/915>.  
Click on "Request Permissions" which will take you to the Copyright Clearance Center's (CCC) Rightslink site.